

5-O-METHYLCYANIDIN 3-GLUCOSIDE FROM LEAVES OF *EGERIA Densa*

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*Egeria densa* (Planch.) St. John has been described as producing a cyanidin diglucoside [1, 2]. However, we have found a new anthocyanin, 5-O-methylcyanidin 3-glucoside, in this species.

Detached leaves were immersed in 0.1 M sucrose soln and incubated at 24–26° for 4–5 days under continuous illumination (ca 4000 lx), whereupon they were tinged with red anthocyanin colour. The red leaves were dried over  $\text{CaCl}_2$  in vacuo and extracted with 0.5% MeOH-HCl. The extract was purified by preparative PC. Bluish-purple needles (10 mg) were separated from the concentrated eluate.

Upon hydrolysis with 6N HCl, the anthocyanin gave glucose and a new aglycone (1) ( $R_f$  0.65 in Forestal, 0.45 in  $\text{HCOOH-HCl-H}_2\text{O}$  (5:2:3)). A bathochromic shift by the addition of  $\text{AlCl}_3$  (5% in EtOH) indicated the presence of an *o*-dihydroxylic B-ring. The absorption spectrum (in 0.01% MeOH-HCl) of 1 showed  $\lambda_{\text{max}}$  272 and 534, and the ratio of  $E_{440}/E_{\text{vis max}}$  was 14%, indicating that it is based on cyanidin.

Upon degradation with 2N NaOH, phloroglucinol monomethylether and protocatechuic acid were detected on TLC [3], suggesting that 1 is a monomethylcyanidin. Its fluorescence in UV-light suggested that the 5-hydroxyl was methylated [4]. This was confirmed by comparing the *Egeria*-aglycone with synthetic 5-O-methylcyanidin, when both compounds were identical in  $R_f$  and in visible and IR spectra.

The original anthocyanin was proved to be 3-monoglucoside of 5-O-methylcyanidin from the results of partial hydrolysis and  $\text{H}_2\text{O}_2$ -degradation. The new glucoside was also detected in a small amount from reddish spring shoots of *Egeria densa* and *Elodea Nuttallii* (Planch.) St. John.

## EXPERIMENTAL

*Plant material.* Collected in early winter from the Botanical garden of Tokyo Kyoiku University.

*$\omega$ -Hydroxy-3,4-diacetoxyacetophenone.* A soln of  $\omega$ -diazo-3,4-diacetoxyacetophenone (97 mg, synthesized from 3,4-diacetoxybenzoic acid chloride and  $\text{CH}_3\text{N}_2$ ) in 50%  $\text{HCOOH}$  was stirred at room temp. for 30 min. After heating at 80° for 10 min.,  $\text{H}_2\text{O}$  was added to the cooled reaction mixture. The EtOAc extract was dried, and evaporated to give pale yellow solids. Recrystallization from EtOH gave colorless needles (83.7 mg): mp 86–87° [5]; IR ( $\nu_{\text{max}}^{\text{KBr}}$ ) 3450 (br.), 1770, 1690, 1603  $\text{cm}^{-1}$ ; NMR (60 MHz,  $\text{CDCl}_3$ )  $\delta$  2.33 (6H, s), 4.86 (2H, s), 7.4–7.8 (3H).

*Cyanidin 5-methyl ether.* HCl was passed into a soln of 2,4-dihydroxy-6-methoxybenzaldehyde (34.5 mg) and  $\omega$ -hydroxy-3,4-diacetoxyacetophenone (51.2 mg) in dry  $\text{Et}_2\text{O}$  (10 ml) on an ice bath for 90 min. The reaction mixture was stood overnight. The solids were thoroughly washed with  $\text{Et}_2\text{O}$ . The EtOH soln was poured into  $\text{Et}_2\text{O}$  and the red crystalline ppt. (24.1 mg) were collected; mp > 300°, VIS ( $\lambda_{\text{max}}^{\text{EtOH}}$ ) 537 nm, ( $\lambda_{\text{max}}^{\text{EtOH}+\text{AlCl}_3}$ ) 571 nm;  $\nu_{\text{max}}^{\text{KBr}}$  3410, 3000 (br.), 1640, 1605, 1580, 1520, 1325, 1270, 1210 and 1038  $\text{cm}^{-1}$ .

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